SYNTHESIS OF 2,5-DISUBSTITUTED 1,3,4-OXADIAZOLES CONTAINING BENZOTHIAZOLYLTHIOL GROUPING

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A series of 2,5-substituted 1,3,4-oxadiazoles containing 2-benzothiazolylthiomethyl grouping has been synthesized by condensing derivatives of (2-benzothiazolylthio)acetic acid with imino ester hydrochlorides and hydrazides of carboxylic acids, by the cyclodehydration of N-acyl-N'-(2-benzothiazolylthioacetyl)hydrazines under the action of POCl₄, and also by the reaction of 2-mercaptobenzothiazole with 2-chloromethyl-1,3,4-oxadiazoles in the presence of sodium methylate.

Keywords: benzothiazole, carboxylic acid hydrazides, carboxylic acid imino ester hydrochlorides, 2-mercaptobenzothiazole, 1.3,4-oxadiazole, condensation.

N-Substituted 2-amino- and 2-alkylthio-1,3,4-oxadiazoles containing benzothiazole fragments in position 5 possess a wide spectrum of biological activity, including anti-inflammatory [1,2], antimicrobial [3], antibacterial [2], and hypotensive [4] activity. There is extremely limited information in the literature [5,6] regarding 2-alkyl(aryl)-5-substituted 1,3,4-oxadiazoles containing benzothiazole fragments.

In continuation of our investigations on the synthesis of heteryl-substituted 1,3,4-oxadiazoles, we give data in the present study on the preparation of 2,5-disubstituted 1,3,4-oxadiazoles containing 2-benzothiazolylthiomethyl grouping. Compounds of this type may be of interest as potentially biologically active substances and also as stabilizers and additives for polymeric material, hydrocarbon fuel, and lubricating oil [10].

Hydrochlorides of carboxylic acid imino esters may serve as convenient synthons in the synthesis of 1,3,4-oxadiazoles [7,10,11]. In the present work the methyl imino ester hydrochlorides of the following acids were used as starting materials: butyric (1a), substituted acetic (1b-f), benzoic (1g), 4-nitro- (1h), and 4-hydroxy-3,5-di(*tert*-butyl)benzoic (1i), β -[4-hydroxy-3,5-di(*tert*-butyl)phenyl]propionic (1j), 5-nitro-2-furancarboxylic (1k), and 3-indolecarboxylic (1l) acids. 2-alkyl(aryl,heteryl)-5-(2-benzothiazolylthiomethyl)-1,3,4-oxadiazoles (3a-m) were formed as a result of the condensation of the imino ester hydrochlorides 1a-l with (2-benzothiazolylthiomethyl)acetic acid hydrazide (2a) (method A).

The best yields of compounds **3a-1** (Table 1) were achieved on boiling the reactants in ethanol or dioxane at a molar ratio of 1 : 2a of 1.2 : 1. The duration of the process depends on the reactivity of the initial imino ester hydrochloride **1a-1**. For example, the formation of compounds **3a-h,j,k** is complete after boiling the reactants for 4-5 h in ethanol. When obtaining 1,3,4-oxadiazoles **3i,1** from imino ester hydrochlorides **1i,1**, which have reduced reactivity due to the effect of bonding of the electron-donating hydroxyaryl or indole substituent with the imino-ester group [11], it was necessary to boil in dioxane for 10-12 h.

We also used the condensation of (2-benzothiazolylthio)acetic acid imino ester dihydrochloride (1f) with hydrazides of various carboxylic acids (2b-f) in the synthesis of 1,3,4-oxadiazoles 3g,h,j-l (method B). The reaction was carried out by boiling the reactants (molar ratio of 1f : 2 was 1.25 : 1) in ethanol for several hours. The corresponding products 3g,h,j-l were formed in 65-76% yields. In addition, compounds 3g,h and 2-methyl-

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1a, 3a R = Pr; 1b, 3b R = ClCH₂; 1c, 3c R = EtOOCCH₂; 1d, 3d R = PhCH₂; 1e, 3e R = 4-NO₂C₆H₄CH₂; 1f, 3f R = 2-benzothiazolylthiomethyl; 1g, 2b, 3g, 4b R = Ph; 1h, 2b, 3h, 4c R = 4-NO₂C₆H₄; 1i, 3i R = 4-HO-3,5-(*t*-Bu)₂C₆H₂; 1j, 2d, 3j R = 4-HO-3,5-(*t*-Bu)₂C₆H₂CH₂CH₂; 1k, 2e, 3k R = 5-nitrofur-2-yl; 1l, 2f, 3l R = 3-indolyl; 3m, 4a R = Me

1,3,4-oxadiazole **3i** were obtained by cyclodehydration of the corresponding N-acyl-N'-(2benzothiazolylthioacetyl)hydrazines (**4a-c**) under the action of phosphorus oxychloride (method C) [7,9,12]. The latter were obtained by acylation of hydrazide **2a** with acetic anhydride in an inert solvent at room temperature or with the acid chlorides in pyridine in 78-85% yield. However even brief heating with phosphorus oxychloride leads to strong resinification of the reaction mixture from which the desired products **3g,h,i** were isolated in 38-47% yield. On boiling N,N'-diacylhydrazines **4a-c** in an excess of acetic anhydride for several hours no cyclodehydration to the corresponding 1,3,4-oxadiazoles occurred.

2-Arylthiomethyl-5-R-1,3,4-oxadiazoles may be obtained by the interaction of 2-chloromethyl-5-R-1,3,4-oxadiazoles with arylthiols [13]. We have used this method in the present work for the synthesis of compounds **3f-h,l**. As a result of the condensation of 2-chloromethyl-5-R-1,3,4-oxadiazoles **3b**, **5a-c** with 2-mercaptobenzothiazole in the presence of equimolar quantity of sodium methylate at 0-10°C (0.5 h), the corresponding 1,3,4-oxadiazoles **3f-h,l** were formed in 78-87% yield (method D).



5a R = Ph, **5b** R = 4-NO₂C₆H₄, **5c** R = 3-indolyl

The characteristics of the disubstituted 1,3,4-oxadiazoles **3a-m** synthesized are given in Table 1. The composition and structure of these compounds were confirmed by data of elemental analysis, IR and ¹H NMR spectroscopy. In the IR spectra intense absorption maxima were observed in the ranges of 1600-1615, 1570-1585, and 1460-1490 cm⁻¹ characteristic of the stretching vibrations of the oxadiazole ring [14,15]. The presence of the latter was confirmed by occurrence of absorption bands at 1225-1250 and 1020-1050 cm⁻¹ assigned to the stretching vibrations of the =C-O-C= fragment in 1,3,4-oxadiazoles [16], and of an absorption maximum near 950 cm⁻¹ related to the breathing vibrations of the oxadiazole ring [14,15]. Absorption due to the benzothiazole fragment [14] was also observed for all the compounds considered at 1520-1530, 1430-1445 (stretching vibrations of the ring), 1060-1085, and 735-740 cm⁻¹.

In the 'H NMR spectra of the synthesized compounds the signals of the thiomethyl group protons were displayed as singlets in the range of 3.94-4.27 ppm. The multiplet signals at 7.14-8.14 ppm correspond to the protons of the benzothiazole fragments.

| Empirical | | | Foun | d. ". | | *. J _o uu | R. tsolven | H NVIR sociation à minit [*] . | Yield, ⁿ . |
|---|--|---|---|-------------|-----------------|----------------------|------------------------|---|-----------------------------------|
| formula – C II N S | | H N S | N S | S. | 1 | | system) | | (method) |
| 3 | v | | 2 5 | ~ | | | × | 6 · · · · · · · · · · · · · · · · · · · | |
| C ₁ ,H ₁ ,N ₁ OS; <u>53,47</u> <u>4,56</u> <u>14,87</u> <u>21.</u> <u>53,61</u> <u>4,47</u> <u>14,73</u> <u>21.</u> | <u>53.47</u> <u>4.56</u> <u>14.87</u> <u>21.</u> <u>53.61</u> <u>4.47</u> <u>14.73</u> <u>21.</u> | <u>4.56</u> <u>14.87</u> <u>21.</u> | 14.87 21. | 20 | 18 | 134-136 | 0.57 (a) | 1.10 (3H, s. CH5): 1.33-1.44 (2H, m. CH5): 2.24 (2H, i, CH <u>5CH5</u>): 3.94 (2H, s. CH5): 7.38-7.97 (4H, m. H5) | 78 (A) |
| C.,H.CIN.OS ₂ <u>44.51</u> <u>2.77</u> <u>14.01</u> <u>2.1</u> <u>2.69</u> <u>14.12</u> <u>21</u> | <u>44.51</u> <u>2.77</u> <u>14.01</u> <u>21.</u> <u>14.37</u> <u>2.69</u> <u>14.12</u> <u>21.</u> | <u>2.69</u> <u>14.12</u> <u>21.69</u> | 17 01 01 01 01 01 01 01 01 01 01 01 01 01 | | 212 | 110 113 | 0.74 (b) | 2.87 (2H, s, CH;CH; 4.12 (2H, s, CH ₅ S); 7.52-8.04 (4H, m, H _v) | 72 (A) |
| C.J.H.N.O.S. 50.03 3.97 12.66 18.5 | <u>50.03</u> <u>3.97</u> <u>12.66</u> <u>18.5</u> <u>50.15</u> <u>3.88</u> <u>12.54</u> <u>19.1</u> | <u>3.87</u> <u>3.88</u> <u>12.54</u> <u>19.1</u> | 12.66 12.54 19.1 | <u>1.61</u> | 10 | ÖÏ | (a) (a) | 1.37 (3H, t, J = 6 Hz, CH ₃ : 3.18 (2H, s, CH ₃); 3.84 (2H, q, J = 6 Hz, CH ₅ O); 4.22 (2H, s, CH ₅ S); 7.14-7.55 (4H, m, H ₃) | 67 (A) |
| C ₁ -H ₁ ,N,OS ₂ 60.06 3.94 12.56 18. 60.18 3.83 12.39 18. | 60.16 3.94 1.2.56 18. 60.18 3.83 12.39 18. | 3.94 12.56 18. 3.83 12.39 18. | 12.56 12.39 18. | <u>x x</u> | <u>27</u> 88 | 44 CX | ()' 1 5 (c) | 3.34 (2H, s, PhCH ₂): 4.07 (2H, s, CH ₂ S): 6.85-7.30 (9H, m, H ₃) | (Y) 1 8 |
| C.H.N.O.S. 53.25 3.04 14.71 16 53.12 3.12 14.58 16 | 53.25 3.04 14.71 16 53.12 3.12 14.58 16 | <u>3.12</u> <u>3.12</u> <u>14.58</u> <u>16</u> | 14.71 14.58 16. | 20 | <u>6</u> | 195 197 | 0.52 (b) | 3.50 (2H, s. ArCH5); 4.21 (2H, s. CH5S); 7.04-7.59 (8H, m. Hx,) | 77 (A) |
| C ₁ ,H ₁ ,N ₁ OS ₁ 50,37 2.92 12.84 2 | 50,37 2,92 12,92 30 13,08 13,08 13,08 30 | <u>2.80</u> <u>13.08</u> | 13.08 13.08 | πir. | 10.0 | 176-178 | 0.62 (c) | 4.24 (4H, s, CH ₅ S): 7.59-8.08 (8H, m, H _w) | 81 (A) 78 (D) |
| $\left \begin{array}{c} C_{1,1}I_{1,1}N_{1,1}OS_{2} \\ \hline S9,08 \\ \hline S9,08 \\ \hline 3.38 \\ \hline 3.38 \\ \hline 12.92 \\ \hline 1 \\ \hline 1 \\ \hline 2 \\ \hline \end{array} \right $ | 59.08 3.47 13.11 1 59.08 3.38 12.92 1 | 3.47 13.11 1 3.38 12.92 1 | <u>13.11</u> 12.92 | | <u>6.47</u> | 95-97 | ().62 (b) | 4.12 (211. s. CH ₂ S); 6.92-7.37 (911. m, H _w) | 87 (A), 76 (B), 42 (C), 83 (D) |

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| 10 | 83 (A), 72 (B), 47 (C), 87 (D) | (V) +(V) | H): 73 (A). | 82 (A). 74 (B) | 76(A), 65 (B), 80 (D) | 38 (C) |
|------------|--|--|--|--|--|---|
| 6 | 4.20 (2H, s, CH ₅ S): 7.05-7.48 (8H, m, H ₃ ,) | 1.52 (18H, br. s. <i>t</i> -Bu); 3.94 (2H, s. CH ₅ S); 5.18 (1H, s. OH); 7.12 (2H, s. 2-, 6-H ₃); 7.55-8.14 (4H, m, H ₃ ,) | 1.60(118H, br. s. <i>t</i> -Bu); 3.72-4.04 (4H, m, CH5CH ₂): 4.27 (2H, s. CH ₂ S); 5.26 (1H, s. O 6.84 (2H, s. 2-,6-H ₃); 7.28-7.62 (4H, m, H ₃₀ , | 4.02 (H. s. CH5S); 6.58 (1H, d. <i>J</i> = 3.7 Hz, 3-H furan); 6.92 (1H, d. 4-H furan); 7.38-7.84 (4H, m, H _v) | 4.15 (2H. s. CH ₂ S): 7.22-7.78 (9H, m, H _w): 8.14 (1H, br. s. NH) | 2.64 (311, s, CH ₁); 4.04 (2H, s, CH ₂ S); 7.48-7.96 (4H, m, H _A) |
| × | 0.53 (a) | ().75 (a) | (),66 (c) | ().43 (a) | 0.53 (b) | 0.58 (c) |
| 7 | 157-158.5 | 145-146 | Oil : (<i>m</i>) ²⁰ 1.5818) | 204-206 | 961-561 | 79-81 |
| y | <u>17.14</u> 17.30 | <u>14.25</u> 14.13 | <u>13.16</u> 13.30 | <u>17.78</u> | <u>17.33</u> 17.58 | <u>24.47</u> 24.33 |
| 5 | <u>15.13</u> | <u>9.05</u> | <u>8.95</u> 8.73 | <u>15.77</u> 15.55 | <u>15.61</u> 15.38 | <u>16.15</u> 15.97 |
| + | <u>2.83</u> 2.70 | <u>5.96</u> | <u>6.30</u> 6.44 | 2.2.2 | <u>3.38</u> 3.30 | <u>3.50</u> <u>3.42</u> |
| 3 | <u>52.04</u> 51.89 | <u>63.69</u> 63.58 | <u>65.07</u> 64.86 | <u>46.53</u> 46.67 | <u>59.21</u> 59.34 | <u>49.96</u> 50.19 |
| c 1 | C _{iv} H _{in} N ₄ O ₃ S ₂ | C ₂₄ H ₂₅ N ₃ O ₂ S ₂ | C ₃ ,H ₁ ,N ₁ O ₂ S ₂ | C ₁₄ H ₈ N ₄ O ₄ S ₂ | C ₁₈ H ₁₂ N ₁ OS ₂ | C ₁₁ H ₅ N ₅ OS ₂ |
| | ųę. | 31 | į£ | 3k | R | 3m |

TABLE 1 (continued)

* Compounds were recrystallized as follows: **3a**,m from toluene-petroleum ether, 1 : 3; **3b-f,h,l** from dioxane-water, 1 : 1.5; **3g** from acetone-water, 1 : 2.5; **3i** from heptane-dioxane, 3 : 1; **3k** from aqueous DMF. *² The spectra of compounds **3a,d-i,k,l** were recorded in DMSO-d₆ and those of compounds **3b,c,j** in CDCl₃.

| Com- pound | Solvent | ()=() | C=C | ()CN (C=N) | C=O ester | NII |
|---------------|---------------------------------|-------|------|---------------|--------------|-----------|
| 3a | Nujol | 1770 | 1655 | 1580, 1550 | 1740 | 3270-2600 |
| | CH ₂ CI ₂ | 1795 | 1665 | 1590 | 1747 | 3290-2650 |
| 3b | Nujol | 1788 | 1660 | 1590, 1551 | 1730 | 3290-2610 |
| | CH2Cl2 | 1780 | 1657 | 1575 | 1742 | 3300-2670 |
| 4 | Nujol | 1740 | 1650 | (1575) | 1740 | 3320-2610 |
| | CHCh | 1745 | 1640 | (1590) | 1740 | 3260-2860 |

TABLE 2. The IR Spectral Characteristics of the Synthesized Compounds 3a,b and $4, v, cm^{-1}$

EXPERIMENTAL

The IR spectra were taken on a Bruker IFS 48 instrument in KBr disks, nujoi suspensions, or in thin films. The 'H NMR spectra were recorded on a Bruker WP 250 spectrometer, internal standard was TMS. A check on the progress of reactions and the purity of the compounds obtained was made with the aid of TLC on Al_iO_i of Brockmann activity grade III in the solvent systems a) benzene–ethanol, 20 : 1; b) benzene–ethanol, 10 : 1; and c) CCl₄–acetone, 20 : 1. Visualization was with iodine vapor.

The initial methyl imino ester hydrochlorides of butyric (1a) [18], phenylacetic (1d) [19], 4-nitrophenylacetic (1e) [19], benzoic (1g) [17], 4-nitrobenzoic (1h) [19], 4-hydroxy-3,5-di(*tert*-butyl)benzoic (1i) [20], β -[4-hydroxy-3,5-di(*tert*-butyl)phenyl]propionic (1j) [20], 5-nitro-2-furancarboxylic (1k) [21], and 3-indolecarboxylic (1l) [22] acids, hydrazides of (2-benzothiazolylthio)acetic (2a) [2], β -[4-hydroxy-3,5-di(*tert*butyl)phenyl]propionic (2d) [23], and 3-indolecarboxylic (2f) [7] acids, and also 2-chloromethyl-1,3,4-oxadiazoles 5a [24], 5b [24], and 5c [25] were obtained by known methods (citations are given above for each compound).

(2-Benzothiazolylthio)acetic Acid Methyl Imino Ester Dihydrochloride (1f). A stream of dry gaseous HCl was passed for 2.5 h into stirred solution of (2-benzothiazolylthio)acetonitrile (8.24 g, 0.04 mol) and absolute methanol (3.2 g, 0.1 mol) in anhydrous 1,2-dimethoxyethane (150 ml) at 0-5°C. The reaction mixture was maintained at 20°C for 24 h, anhydrous ether (150 ml) was poured in, and the mixture cooled to -5°C. The precipitated solid was filtered off, washed on the filter with anhydrous ether to neutral reaction, and dried in vacuum over KOH. Imino ester dihydrochloride 1f (9.44 g, 86%) was obtained; mp 173-175°C (decomp.). IR spectrum, cm⁻¹: 740, 3120, 1675, 1525, 1420, 1030. Found, %: Cl 23.04; N 3.91. C₁₀H₁₀N₂OS₂·2HCl. Calculated, %: Cl 22.82; N 9.00.

2,5-Disubstituted 1,3,4-Oxadiazoles (3a-I). A. Mixture of imino ester hydrochloride **1a-I** (12.0 mmol) and hydrazide **2a** (2.39 g, 10.0 mmol) in absolute ethanol (45 ml) was boiled with stirring for 5 h (when obtaining compounds **3i,I** the mixture was boiled for 12 h in 45 ml of anhydrous dioxane). The reaction mixture was evaporated to dryness at reduced pressure. In the case of compounds **1a,e-i,k,I** the residue was crystallized from a suitable solvent (see Table 2), and on making compounds **3b-d,j** it was chromatographed on column with Al₂O₄ (h = 80 cm, d = 4.5 cm) eluting with mixture of benzene-methanol, 15 : 1. After removing the solvent, 1,3,4-oxadiazoles **3c,j** were obtained as viscous dark yellow uncrystallizable oils.

5-(2-Benzothiazolylthiomethyl)-2-phenyl-1,3,4-oxadiazole (3g). B. Mixture of imino ester dihydrochloride **1f** (3.87 g, 12.5 mmol) and hydrazide **2b** (1.48 g, 10.0 mmol) in absolute ethanol (40 ml) was boiled with stirring for 4-5 h until the initial hydrazide **2b** had disappeared from the reaction mixture (check by TLC). The solvent was removed at reduced pressure, the residue crystallized from mixture of acetone–water, 1 : 1.25, and 1,3,4-oxadiazole **3g** was obtained.

1,3,4-Oxadiazoles **3h,j-l** were synthesized similarly from imino ester **1f** and hydrazides **2c-f** respectively.

N-Acetyl-N'-(2-benzothiazolylthioacetyl)hydrazine (4a). Acetic anhydride (3.6 g, 35 mmol) was added during 0.5 h to stirred mixture of hydrazide **2a** (7.17 g, 30.0 mmol) and 2-propanol (70 ml) at 20°C. The reaction mixture was stirred at 20°C for 2 h then ice water (50 ml) was added dropwise. The precipitated solid was filtered off, dried, and crystallized from ethanol. Hydrazine **4a** (7.16 g, 85%) was obtained; mp 168-170°C, R_t 0.67

(system b). IR spectrum: 3200-3320, 3050, 2910, 1665, 1640, 1530, 1440, 1260, 1240, 1150, 890, 825, 735 cm⁻¹. ¹H NMR spectrum (DMSO-d_a): 2.74 (3H, s, Me); 4.18 (2H, s, CH₂S); 7.37-7.84 (4H, m, H_{Ar}); 8.18 ppm (2H, br. s, NH). Found, %: C 47.14; H 3.82; N 15.19; S 22.54. $C_{11}H_{11}N_1O_2S_2$. Calculated, %: C 46.97; H 3.91; N 14.95; S 22.78.

N-Benzoyl-N'-(2-benzothiazolylthioacetyl)hydrazine (4b). Benzoyl chloride (2.81 g, 20.0 mmol) was added in portions to stirred solution of hydrazide **2a** (4.78 g, 20.0 mmol) in anhydrous pyridine (50 ml). The reaction mixture was boiled with stirring for 2 h, cooled to 20°C, and poured into ice water (200 ml). The precipitated solid was filtered off, washed on the filter with water, dried, and crystallized from 1-butanol. Hydrazine **4b** (5.35 g, 78%) was obtained; mp 154-156°C, R_i 0.43 (system b). ¹H NMR spectrum (DMSO-d₆): 4.04 (2H, s, CH₂); 6.74-7.22 (9H, m, H_{Ai}); 8.02 ppm (2H, br. s, NH). Found, %: C 56.16; H 3.63; N 12.02; S 18.85. C₁₆H₁₄N₃O₂S₂. Calculated, %: C 55.98; H 3.79; N 12.24; S 18.66.

N-(4-Nitrobenzoyl)-N'-(2-benzothiazolylthioacetyl)-hydrazine (4b) was obtained analogously from 4-nitrobenzoyl chloride. Yield 81%; mp 166-168°C (dioxane–water, 1 : 1); R_1 0.62 (system b). Found, %: C 49.35; H 2.90; N 14.61; S 16.62. C₁₀H₁₂N₄O₄S₂. Calculated, %: C 49.48; H 3.09; N 14.43; S 16.49.

5-(2-Benzothiazolylthiomethyl)-2-methyl-1,3,4-oxadiazole (3m). C. Mixture of hydrazine **4a** (4.20 g, 15.0 mmol) and POCl₃ (30 ml) was stirred at 80-85°C for 0.5 h. The reaction mixture was cooled to 0°C, poured onto ice (200 g), and the mixture neutralized to pH 7.5 with aqueous ammonia solution. The dark oil which precipitated was extracted with dichloromethane (3×20 ml), the extract washed with water, dried over Na₃SO₄, and evaporated to dryness under reduced pressure. The residue was crystallized from toluene–petroleum ether, 1 : 3, and 1,3,4-oxadiazole **3m** was obtained.

1,3,4-Oxadiazoles **3g,h** were synthesized analogously from N,N'-diacylhydrazines **4b,c** respectively.

5-(2-Benzothiazolylthiomethyl)-2-(4-nitrophenyl)-1,3,4-oxadiazole (3h). D. 2-Mercaptobenzothiazole (2.0 g, 12.0 mmol) was added in portions to stirred solution of sodium methylate obtained from sodium (0.2 g, 14.0 g-atom) in absolute methanol (65 ml). The reaction mixture was stirred at 20° C for 0.5 h, cooled to 0° C, and 2-chloromethyl-1,3,4-oxadiazole **5b** (2.87 g, 12.0 mmol) was added in portions. The reaction mixture was then stirred at 0-10°C for 0.5 h, left for 1 h at 20°C, and evaporated to dryness at reduced pressure. The residue was crystallized from dioxane–water, 1 : 1.5, and 1,3,4-oxadiazole **3h** was obtained.

1,3,4-Oxadiazoles **3f,g,l** were synthesized analogously from 2-chloromethyl-1,3,4-oxadiazoles **3b,5a,c** respectively.

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